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Photochemical reactions of 1-methyl-4,6-diaryl-2(1H)pyrimidinones 1a-b in the presence of thiols 2 are described. Irradiation of 1-methyl-4,6-diaryl-2(1H)-pyrimidinones 1a-b in benzene in the presence of thiols 2 gave the unexpected 2:1-adducts, 3-methyl-4,6-diaryl-5-aralkylthio-6-(1'-methyl-4',6'-diaryldihydro-pyrimidin-2-on)yl-1,3-diazabicyclo[2.2.0]hexan-2-ones 3-6, of 1 and 2, whereas irradiation of 1a-b alone in benzene resulted in recovery of the unchanged 1a-b.

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In course of our studies on the photochemical reactivities of cyclic conjugated nitrogen-carbonyl systems such as 2(1H)-pyrimidinones [1-9] and 2(1H)-pyrazinones [10-14], we found that upon irradiation 2(1H)-pyrimidinones undergo various types of photochemical reactions depending on the substituents in the 1-, 4-, and 6-positions [1-9]. For examples, 1,4,6-trisubstituted 2(1H)pyrimidinones undergo valence isomerization to yield 1,3-diazabicyclo[2.2.0]hex-5ene-2-ones (Dewar pyrimidinones) [2-3]. Irradiation of N-aryl-2(1H)-pyrimidinones gave arylimine derivatives [4-6], which are produced by initial Norrish-type I cleavage of ArN-CO bond of 2(1H)-pyrimidinones; in alcohols 1-alkoxycarbonylamino-3-arylimino-1-propenes derivatives [4-5] and in the presence of amines 1-(N'-alkylureido)-3arylimino-1-propenes [6] were produced. These are analogous to conjugated cyclohexanones [15] and 2-pyrones [16]. Inter- and intramolecular hydrogen abstraction reactions by nitrogen of the imino group of 2(1H)-pyrimidinones were also observed [7-9]. In this context, we report here the photochemical reactions of 1-methyl-4,6-diaryl-2(1H)-pyrimidinones 1 in the presence of thiols 2, which formed the unexpected 2:1 adducts 3-6 of 1 and 2.

Photolysis of 1-methyl-4,6-diaryl-2(1H)-pyrimidinones 1 alone in benzene or methanol resulted in recovery of the starting materials 1 [7]. However, irradiation of 1-methyl-4,6-diphenyl-2(1H)-pyrimidinone **1a** in benzene in the presence of ethanethiol 2a in a Pyrex vessel with a high-pressure mercury lamp under argon for 15 hours at room temperature gave a complex mixture of products from which a white crystalline solid, mp 175-176°, was isolated when the reaction mixture was purified through a silica gel column chromatography. This compound had the molecular formula C₃₆H₃₄N₄O₂S, which was equivalent to the 2:1 adduct 3 of the 2(1H)-pyrimidinone 1a and ethanethiol 2a. The ir spectrum of 3 showed absorptions at 3400 and 3300 cm⁻¹ due to amino group, 1785 cm⁻¹ due to typical of Dewar pyrimidinones [2,17], and 1680 and 1660 cm-1 due to amide carbonyls. 1H nmr spectrum of 3 displayed a singlet at δ 4.92 (1H) and a broad singlet at δ 5.23 (2H, 1H disappeared by treatment of deuterium oxide), which were assigned to methine at C-5 and olefinic at C-5' and amino protons, respectively. ¹³C nmr spectrum of 3 showed a olefinic carbon at δ 105.1 (d, C-5'), three quaternary carbons at δ 68.5 (s, C-6 or C-6'), 77.7 (s, C-6' or C-6) and 80.9 (s, C-4), and two ureido carbonyl carbons at δ 154.2 (s), and 159.9 (s). Thus, the photoproduct was characterized as 3-methyl-4,6-diphenyl-5-ethylthio-6-(1'-methyl-4',6'-diphenyldihydropyrimidin-2-on)yl-1,3-diazabicyclo[2.2.0]hexan-2-one 3. Similarly, irradiation of 1-methyl-4,6-diaryl-2(1*H*)pyrimidinones 1a-b in benzene in the presence of thiols 2 under the same conditions as described above gave the corresponding 2:1 adducts 4-6 of 1a-b and 2 in 19-30% yields. The structure of 2:1 adducts 4-6 was elucidated on the basis of their spectroscopic properties and elemental analyses (see Experimental section).

On the other hand, irradiation of 1-phenyl-4,6-dimethyl-2(1H)pyrimidinone 1c in benzene in the presence of thiols 2a,c gave the 1:1 adduct, 5-aralkylthio-4,6-dimethyl-3-phenyl-1,3-diazabicyclo[2.2.0]hexan-2-ones 7-8 [4] of Dewar pyrimidinone, which was formed by photochemical electrocyclization of 1c [2], and thiols 2. Although the mechanism for the formation of 2:1 adducts 3-6 of the 2(1H)-pyrimidinones 1 and thiols 2 is unclear at present, we postulate the reaction path as shown in Scheme 2 that involves $[2\pi + 2\pi]$ photodimerization between N3-C4 and C5-C6 bonds leading to cyclobutane C, which produces the final product by the addition of thiol followed by ring opening of cyclobutane.

EXPERIMENTAL

Melting points were measured with Yanaco micro melting point apparatus (MP-3J) and are uncorrected. The ir spectra were recorded on a Hitachi 260-30 spectrophotometer (in potassium bromide and are in cm⁻¹). The uv spectra were recorded on a JASCO UVDEC-505 spectrophotometer (in ethanol and in nm). 1 H and 13 C nmr spectra were run on a JEOL FX-100 spectrometer (100 MHZ) in deuteriochloroform using TMS as an internal standard (δ in ppm, J in Hz).

General Procedure for the Photochemical Reactions of the 2(1H)-Pyrimidinones 1 in the Presence of Thiols 2.

A solution of the pyrimidinone 1 (1 mmole) in benzene (50 ml) in the presence of an excess of the thiol (ca 0.5 ml) was irradiated in a Pyrex vessel with a high-pressure mercury lamp (300 W) under an argon atmosphere for 15 hours at room temperature. After removal of the solvent, the residue was chromatographed on a silica gel column with benzene/ethyl acetate (19:1) as eluant to give 5-aralkylthio-6-(pyrimidin-2'-on)yl-1,3-diazabicyclo-[2.2.0]hexan-2-ones 3-6.

3-Methyl-4,6-diphenyl-5-ethylthio-6-(1'-methyl-4',6'-diphenyl-dihydropyrimidin-2'-on)yl-1,3-diazabicyclo[2.2.0]hexan-2-one (3).

This compound had mp 175-176°; uv: λ 274 (ϵ 6.2 x 10³); ir: v 3400, 3300 (NH), 1785, 1680, 1660 (CO); ¹H nmr: δ 1.49 (t, 3H, J = 7.3, CH₃), 2.51 (s, 3H, NMe), 2.79 (s, 3H, NMe), 2.89 (q, 2H, J = 7.3, CH₂), 4.92 (s, 1H, CH), 5.23 (br s, 2H, NH and =CH, 1H diasppeared by addition of deuterium oxide), 6.45-6.55 (2H, m, aromatic), 6.99-7.35 (16H, m, aromatic), 7.48-7.61 (m, 2H, aromatic); ¹³C nmr: δ 14.7 (q, CH₃), 27.8 (q, NMe), 28.5 (t, CH₂), 32.3 (q, NMe), 55.4 (d, CH), 65.8 (s, quaternary C), 77.7 (s, quaternary C), 80.9 (s, quaternary C), 105.1 (d, =CH), 154.2 (s, CO), 159.9 (s, CO) in addition to aromatic carbon peaks.

Anal. Calcd. for $C_{36}H_{34}N_4O_2S$ (586.742): C, 73.69; H, 5.84; N, 9.55. Found: C, 73.31; H, 5.81; N, 9.46.

3-Methyl-4,6-diphenyl-5-propylthio-6-(1'-methyl-4',6'-diphenyl-dihydropyrimidin-2'-on)yl-1,3-diazabicyclo[2.2.0]hexan-2-one (4).

This compound had mp 168-169°; uv: λ 275 (ϵ 5.9 x 10³); ir: v 3420, 3300 (NH), 1790, 1680, 1660 (CO); 1H nmr: δ 1.16 (t, 3H, J = 7.3, CH₃), 1.75-1.96 (m, 2H, CH₂), 2.51 (s, 3H, NMe), 2.79 (s, 3H, NMe), 2.84 (t, 2H, J = 7.8, CH₂), 4.98 (s, 1H, CH), 5,23 (br s, 2H, NH and =CH), 6.44-6.53 (m, 2H, aroamtic), 7.02-7.36 (m, 16H, aromatic), 7.42-7.60 (m, 2H, aromatic); 13 C nmr: δ 13.5 (q, CH₃), 23.0 (t, CH₂), 27.8 (q, NMe), 32.3 (q, NMe), 36.5 (t, CH₂), 55.7 (d, CH), 65.7 (s, quaternary C), 77.7 (s, quaternary C), 80.9 (s, quaternary C), 105.0 (d, =CH), 154.0 (s, CO), 159.9 (s, CO) in addition to aromatic carbon peaks.

Anal. Calcd. for C₃₇H₃₆N₄O₂S (600.768): C, 73.97; H, 6.04; N, 9.32. Found: C, 73.58; H, 6.00; N, 9.19.

3-Methyl-4,6-diphenyl-5-phenylthio-6-(1'-methyl-4',6'-diphenyldihydropyrimidin-2'-on)yl-1,3-diazabicycl [2.2.0]-hexan-2-one (5).

This compound had mp 167.5-168.5°; uv: λ 258 (ϵ 1.08 x 10⁴); ir: 3430, 3300 (NH), 1790, 1680, 1660 (sh) (CO). 1 H nmr: δ 2.58 (s, 3H, NMe), 2.77 (s, 3H, NMe), 5.23 (br s, 2H, NH and =CH), 5.42 (s, 1H, CH), 6.44-6.53 (m, 2H, aromatic), 7.06-7.68 (m, 23H, aromatic).

Anal. Calcd. for C₄₀H₃₄N₄O₂S (634.782): C, 75.68; H, 5.40; N, 8.64. Found: C, 75.36; H, 5.29; N, 8.73.

3-Methyl-4,6-di-tolyl-5-ethylthio-6-(1'-methyl-4',6'-ditolyldihydropyrimidin-2'-on)yl-1,3-diazabicyclo[2.2.0]-hexan-2-one (6).

This compound had mp 155-156°; uv: λ 220 (sh, ϵ 4.07 x 10⁴), 272 (ϵ 6.7 x 10³); ir: v 3420, 3260 (NH), 1785, 1675, 1660 (CO).

¹H nmr: δ 1.47 (t, 3H, J = 7.8, CH₃), 2.26 (s, 3H, Me), 2.28 (s, 3H, Me), 2.32 (s, 3H, 2 x Me), 2.50 (s, 3H, NMe), 2.78 (s, 3H, NMe), 2.85 (q, 2H, J = 7.8, CH₂), 4.87 (s, 1H, CH), 5.18 (br s, 2H, NH and =CH), 6.47 (d, 2H, J = 7.8, aromatic), 6.87-7.42 (m, 14H, aromatic).

Anal. Calcd. for C₄₀H₄₂N₄O₂S (642.846): C, 74.73; H, 6.58; N, 8.71. Found: C, 74.55; H, 6.59; N, 8.56.

REFERENCES AND NOTES

- [1] T. Nishio and C. Kashima, in Reviews on Heteroatom Chemistry, Vol 13, S. Oae, ed, MYU, Tokyo, 1995, p 149.
- [2] T. Nishio, A. Kato. Y. Omote, and C. Kashima, *Tetrahedron Letters*, 1543 (1978); T. Nishio, A. Kato, C, Kashima, and Y. Omote, *J. Chem. Soc.*, *Perkin Trans. 1*, 607 (1980).
- [3] T. Nishio, K. Katahira, and Y. Omote, Tetrahedron Letters, 21, 2825 (1980); T. Nishio and Y. Omote, J. Chem. Soc., Perkin Trans. 1, 1773 (1983).
- [4] T. Nishio K. Katahira, A. Kato, C. Kashima, and Y. Omote, *Tetrahedron Letters*, 4211 (1979); T. Nishio, K. Katahira, and Y. Omote J. Chem. Soc., Perkin Trans. 1, 943 (1981).
 - [5] T. Nishio, Liebigs Ann. Chem., 71 (1992).
- [6] T. Nishio, and Y. Omote, J. Chem. Soc., Perkin Trans. 1, 239 (1984).
- [7] T. Nishio, K. Katahira, and Y. Omote, *Chem. Letters*, 1675 (1982); T. Nishio, and Y. Omote, *J. Chem. Soc.*, *Perkin Trans.* 1, 957 (1988).
- [8] T. Nishio, S. Kameyama, and Y. Omote, J. Chem. Soc., Perkin Trans. 1, 1147 (1986).
- [9] T. Nishio, M. Kato, and C. Kashima, Liebigs Ann. Chem., 611 (1989).
- [10] T. Nishio, N. Nakajima, and Y. Omote, *Tetrahedron Letters*, 22, 753 (1981); T. Nishio, N. Nakajima, M. Kondo, Y. Omote, and M. Kaftory, *J. Chem. Soc.*, *Perkin Trans. 1*, 391 (1984).
- [11] T. Nishio, M. Kondo, and Y. Omote, J. Chem. Soc., Perkin Trans. 1, 2497 (1985).
- [12] T. Nishio, T. Nishiyama, and Y. Omote, Tetrahedron Letters, 27, 5637 (1986); T. Nishio T. Nishiyama, and Y. Omote, Tetrahedron, 47, 2979 (1991).
- [13] T. Nishio, N. Tokunaga, M. Kondo, and Y. Omote, J. Chem. Soc., Perkin Trans. 1, 2921 (1988).
- [14] T. Nishio, M. Kondo, and Y. Omote, Helv. Chim. Acta, 74, 225 (1991).
- [15] N. J. Turro, in Modern Molecular Photochemistry, N. J. Turro, ed, Benjamin/ Cumming, Menlo Park 1978, p. 512.
- [16] H. Javaheripour and D. C. Necker, J. Org. Chem., 42, 1844 (1977).
- [17] J-S. Taylor and M. P. Chors, J. Am. Chem. Soc., 109, 2834 (1987).